# FORMATION OF ORTHOESTERS IN THE SHARPLESS ASYMMETRIC EPOXIDATION : HEMISYNTHESIS OF LABDANES.

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(Received in UK 3 January 1990)

Summary: The orthoesters, 4a and 4b were obtained from treatment of methyl-7,13E-labdadien-17-oate, 3, under Sharpless asymmetric epoxidation conditions. The hydrolysis followed by dehydration of the orthoesters led to the synthesis of two new diterpenic acids isolated from Halimium viscosum: 14R,15-dihydroxy-7,13(16)-labdadien-17-oic and 14S,15-dihydroxy-7,13(16)-labdadien-17-oic.

#### Introduction

Several acidic diterpenes have been isolated as their methyl esters from the acid fraction of an extract of *Halimium viscosum* (Valparaiso)<sup>1,2,3</sup>, together with the two unsaturated diols 1 and 2, the two epimers at C-14 of methyl 14,15-dihydroxy-7,13(16)labdadien-17-oate.

The stereochemistry at C-14 was determined by asymmetric hemisynthesis of 1 and 2 from methyl 15-hydroxy-7,13 E-labdadien-17-oate 3, the major product isolated from the plant.

Treatment of 3 under Sharpless asymmetric epoxidation conditions<sup>4</sup> afforded orthoesters instead of epoxides. Hydrolysis and acetylation of 5a and 5b gave the two lactones 6a and 6b and the compounds 7a and 7b respectively, as predicted by Deslongchamps<sup>5,6,7</sup>.

Dehydration of 7a and 7b led to the natural products 1 and 2, the C-14 stereochemistry of which was further corroborated by circular dichroism.

#### Results and discussion

The most polar components of the acidic fraction were the epimeric unsaturated dihydroxy compounds 1 and 2 (IR 3480, 1730, 1650 cm<sup>-1</sup>).

In the <sup>1</sup>H NMR spectrum signals corresponding to the -CH=C-COOMe group were observed (6.7 ppm, 1H, m; 3.7 ppm, 3H,s) together with three methyl singlets (0.91, 0.87 and 0.82 ppm) which were similar to those observed for the rest of the components of the acidic fraction.

The difference between 1 and 2 was manifested only in slight variations in the signals due to the CH<sub>2</sub>=C-CHOH-CH<sub>2</sub>OH group of the side chain, from which it was deduced that these compounds were the C-14 epimers of methyl 14(R/S), 15-dihydroxy-7,13(16)-labdadien-17-oate.

1 and 2 were synthesised from the major component of the acidic fraction of the extract of H. viscosum (Valparaiso)  $3^1$ .

The retrohemisynthetic plan for 1 and 2 is shown in scheme I.

Enantioselective epoxidation following the procedure of Sharpless<sup>4</sup> should lead to 10a and 10b. Transformation of these to give 1 and 2, respectively, could then be accomplished by direct ring opening of the oxirane with a base<sup>8</sup> or alternatively by dehydration of the triols arising from acid hydrolysis using HClO<sub>4</sub>/DMF<sup>9</sup>.

#### Synthesis of 1.

Reaction of 3 under Sharpless asymmetric epoxidation conditions using L(+) diethyl tartrate gave 4a (scheme II).

The conclusion that 4a was not the desired oxirane product of the Sharpless reaction could be drawn immediately on examination of its spectroscopic properties (there were no signals due to protons on an oxirane ring in the <sup>1</sup>H NMR spectum).

The <sup>1</sup>H NMR spectum of a mixture of epoxides 10a and 10b, obtained by reaction of 3 with mCPBA, showed a signal due to the oxirane protons at 2.9 ppm (1H, m).

In the <sup>1</sup>H NMR spectum of 4a signals were observed at: 6.43 ppm (1H, m, H-7); 3.98 (1H,dd), 3.88 (1H, dd) and 3.68 ppm (1H, dd) due to the ABX system of a  $-O\underline{H}_2C-C\underline{H}-O-$  group; 3.42 ppm (3H,s)due to a methoxy group (not a methyl ester) and four methyl singlets, one of which at 1.43 ppm was deshielded by a neighboring oxygenated function.

In the <sup>13</sup>C-NMR signals due to 21 carbons were observed, one of which, a quaternary carbon at 117.91 ppm, suggested the presence of an orthoester.

The position of the free hydroxyl group was determined by acetylation of 4a to give 5a, which showed a signal at 4.19 ppm (2H, m) in its <sup>1</sup>H NMR spectrum, due to H-15.

The orthoester structures of 4a and 5a depicted in the figures were deduced from all these data.

A general procedure for the preparation of orthoesters consists of reaction of an ester with an epoxide under Lewis acid conditions <sup>10</sup>.

a) Ti(iPrO)<sub>4</sub>, L(+)DET, tBuOOH,CH<sub>2</sub>Cl<sub>2</sub>; b) Ac<sub>2</sub>O/Py.; c) HCl, Me<sub>2</sub>CO; d) POCl<sub>3</sub>, Py.; e) NaOH/MeOH (10%). scheme II

In this work it is clear that, due to the presence of the Lewis acid Ti(iPrO)<sub>4</sub> in the reaction mixture, the epoxide initially formed in the Sharpless epoxidation undergoes further intramolecular reaction with the C-17 methoxycarbonyl group <sup>11</sup>.

The hydrolysis of geranyl acetate 2,3 epoxide<sup>9</sup> with  $HClO_4/DMF$  gives, after hydrolysis of the formates, a diol in which the configuration at C-3 has been inverted. This reaction goes via a cyclic ortho amide.

In the reaction described here the opening of the oxirane ring occurs in the same way, but under these conditions the orthoester formed is stable and appears as a reaction product. Inversion takes place at C-13 because, as will be shown later, C-14 retains the stereochemistry predicted by Sharpless.

Hydrolysis of 5a with hydrochloric acid in acetone at pH 3 followed by acetylation of the crude mixture gave, after chromatography, the lactone 6a (IR 1710 cm<sup>-1</sup>) and the acetoxy ester 7a (IR 3500, 1720 cm<sup>-1</sup>; <sup>1</sup>H-NMR 3.69 ppm, 3H, s). The result of this hydrolysis is in accordance with the work of Deslongchamps<sup>5,6,7</sup>.

Stereoelectronic effects in 5a determine that the initial bond breaking process occur at either the  $C_{17}$ -O- $C_{13}$  bond or at the  $C_{17}$ -O- $C_{14}$  bond, but not at the  $C_{17}$ -O-Me bond; this results in the formation of a hemiorthoester, a tetrahedral intermediate, which may adopt various conformations from which, after acetylation, either the lactone 6a or the acetoxy-ester 7a are obtained.

The disubstituted olefin 8a (IR: 3080, 1640 and 890 cm<sup>-1</sup>; <sup>1</sup>H-NMR 5.07 ppm (1H,s), 4.97 ppm (1H,s); <sup>13</sup>C-NMR 122.45 ppm C-16) and trisubstituted olefin 9a (<sup>1</sup>H-NMR 5.49 ppm (1H,t), 1.63 ppm (3H,s)) were obtained from the reaction of 7a with phosphorus oxychloride.

Basic hydrolysis (NaOH/MeOH 10%, 6 h, r.t.) of 8a deprotected the hydroxyl group to give the dihydroxy compound 1 directly.

The absolute configuration at C-14 of 1 was confirmed by circular dichroism. Thus, when 1 was dissolved in hexane and  $Pr(dpm)_3$  added, a positive Cotton effect ( $\Delta \epsilon_{313} = +1.2$ ) was observed in the C.D. curve, confirming that the configuration at C-14 was (R)<sup>12</sup>, the desired stereochemistry.

Therefore 1 is Methyl 14 R, 15-dihydroxy-7,13(16)-labdadien-17-oate.

### Synthesis of 2.

2 was obtained following a parallel synthesis to that of 1. In this instance, the use of D(-) DET in the Sharpless epoxidation led to the orthoester 4b, which gave 5b on acetylation, (scheme III).

a) TI(iPrO) $_4$ , D(-)DET, tBuOOH,CH $_2$ Cl $_2$ : b) Ac $_2$ O/Py.; c) HCl, MeOH/H $_2$ O;d)K $_2$ CO $_3$ , ICH $_3$ ;

e) POCl<sub>3</sub>,Py.; f) NaOH/MeOH (10%)

scheme III

Treatment of 5b under the same conditions as previously applied to 5a (acid hydrolysis using HCl/acetone at pH 3 followed by acetylation) led only to a poor yield of the lactone 6b and trace quantities of the ester 7b. When 5b was subjected to hydrolysis with HCl/MeOH/H<sub>2</sub>O at pH 1, the ratio of ester to lactone products was inverted, but once again high yields were not obtained.

The acetoxy ester 7b was obtained in good yield by  $K_2CO_3/MeOH$  hydrolysis of the  $\varepsilon$ -lactone 6b, followed by heating under reflux with methyl iodide<sup>13</sup> and acetylation.

Dehydration of 7b with POCl<sub>3</sub> produced a mixture of olefins 8b and 9b, the spectra of which were very similar to those of the corresponding epimers 8a and 9a, respectively.

Alkaline hydrolysis of 8b (10% NaOH/MeOH, 7 h, r.t.) gave 2.

The circular dichrosim of 2, determined as before in hexane in the presence of  $Pr(dpm)_3$ , showed a negative Cotton effect ( $\Delta \epsilon_{313} = -0.3$ ) in the CD curve, confirming that the C-14 stereochemistry of 2 was  $S^{12}$ .

Thus, 2 is Methyl 14S,15-dihydroxy-7,13(16)-labdadien-17-oate.

TABLE.- 13C Chemical shifts, in ppm. Assignment based on DEPT experiments.

<u>C</u>	1	2	<u>4a</u>	<u>4b</u>	<u>5b</u>	<u>6a</u>	<u>6b</u>	<u>7a</u>	<u>7b</u>	<u>8a</u>	<u>8b</u>	<u>9b</u>
1	39.59	39.50	39.51	39.18	39.20	39.01	38.47	39.32	39.42	39.55	39.48	40.15
2	18.59	18.58	18.97	18.92	18.92	18.66	18.72	18.56	18.60	18.58	18.58	18.61
3	42.12	42.13	42.16	42.21	42.22	41.94	42.04	42.11	42.14	42.16	42.14	42.04
4	32.84	32.84	32.96	32.90	32.89	32.90	33.18	33.06	32.83	32.84	32.84	32.85
5	49.57	49.56	49.64	49.41	49.42	49.00	49.85	49.51	49.56	49.61	49.58	49.51
6	24.11	24.09	23.14	23.19	23.50	24.61	25.42	24.23	24.19	24.04	24.04	25.47
7	137.93	137.71	125.26	125.29	125.62	142.65	144.31	138.86	138.29	137.36	137.39	137.91
8	135.08	135.12	135.61	137.29	136.64	135.08	133.16	134.71	134.85	135.20	135.17	134.22
9	51.01	51.04	52.18	51.62	51.67	50.80	50.51	51.54	51.42	51.02	51.04	50.02
10	37.06	37.01	36.63	36.45	36.46	36.55	36.58	37.07	37.16	37.07	37.01	36.94
11	27.54	27.60	22.21	23.56	23.19	22.41	21.61	21.27	21.44	27.15	27.26	23.84
12	34.53	34.44	34.33	37.84	37.78	37.70	35.53	40.41	40.72	34.44	34.60	129.97
13	149.57	149.56	81.38	83.62	83.73	82.55	81.18	73.11	80.68	145.58	145.49	129.10
14	74.91	75.03	85.71	85.35	81.32	75.24	76.55	76.58	75.54	73.99	73.81	75.97
15	65.78	65.80	62.13	60.85	62.18	62.83	62.80	63.27	63.23	64.69	64.63	64.48

20.87

162.52

33.11

21.97

13.76

171.51

20.72

170.52

20.72

24.20

168.68

33.12

21.18

13.28

170.77

20.72

170.18

20.72

22.47

169.52

32.82

21.81

14.22

51.54

170.85

21.00

170.45

20.81

23.75

169.51

32.83

21.85

14.32

51.42

170.72

20.89

170.23

20.76

112.45

169.95

33.14

21.94

14.32

51.28

171.22

21.01

168.53

20.71

112.50

169.48

33.14

21.93

14.31

51.31

170.58

21.03

169.90

20.74

14.73

169.03

33.33

22.16

13.14

51.30

170.68

20.99

169.90

20.76

## Experimental part

16

17

18

19

20

COO-Me

OOC-Me

OOC-Me

OOC-Me

OOC-Me

О-Ме

110.88

169.63

33.12

21.91

14.34

51.45

110.73

169.64

33.12

21.92

14.31

51.41

25.93

119.71

33.21

22.16

13.99

48.61

23.56

119.27

33.30

22.09

13.65

48.73

23.44

119.57

33.29

22.07

13.63

48.73

170.65

20.83

I.R. spectra were measured as films for oils on a Beckman-33-IR spectrophotometer. The 200 MHz <sup>1</sup>H NMR and 50 MHz <sup>13</sup>C NMR spectra were determined on a Bruker WP-200-SY spectrometer usually in CDCl<sub>3</sub>, except where otherwise indicated. Chemical shifts in ppm are reported as relative to TMS in the <sup>1</sup>H NMR and to CDCl<sub>3</sub> at 77.0 ppm in the <sup>13</sup>C NMR. Mass spectra were recorded on a VG-TS 250 instrument at 70 ev. m/z followed by relative abundance (%) are stated. Optical rotations were performed in a chloroformic solution, except where otherwise indicated, with a digital Perkin-Elmer-241 polarimeter. Microanalysis were performed by the Instituto de Quimica organica de CSIC (Madrid, Spain).

The raw-material 3 was isolated from a n-hexane extract of Halimium viscosum, as reported in reference<sup>1</sup>.

Sharpless reaction of 3: 4a.-A 100 ml, flask equiped with a Teflon-coated magnetic stirring bar is oven-dried, then flushed with nitrogen. The flask is charged with dry dichloromethane (10 ml) and cooled by stirring in a -23° C bath. Then the following liquids are added sequentially via syringe while stirring in the cooling bath: titanium tetraisopropoxide (0.88 ml, 3 mmol), L-(+)-diethyl tartrate [L-(+)-DET, 0.55 ml, 3 mmol]; the mixture is stirred 5 min. before the next addition, 3 (1 g, 3 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) and, finally, t-butyl hydroperoxide (2 ml, 6 mmol). The resulting homogeneous solution is then stored overnight (18 hours) in the freezer at

-20° C in the sealed reaction vessel. Then the flask is placed in a -23° C bath and 10% aqueous tartaric acid solution (10 ml) is added while stirring; the aqueous layer solidifies. After 30 min., the cooling bath is removed and stirring is continued at room temperature until the aqueous layer becomes clear. After separation of the aqueous layer, the organic layer is washed once with water, dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated to give a yellow oil (1.2 g); this is chromatographed on SiO<sub>2</sub> eluting with hexane: AcOEt (80:20) to afford 4a.- (740 mg, 74 %) Colourless oil; ν<sub>max</sub> (film) 3440, 3060, 1625, 1450, 1250, 1180, 1100, 960; δ<sub>H</sub> (CDCl<sub>3</sub>) 6.34 (1H, m, 7-H), 3.98 (1H, dd, J = 7.3, 3.4 Hz, 14-H), 3.88 (1H, dd, J = 11.2, 7.3, 15-H<sub>A</sub>), 3.68 (1H, dd, J = 11.2, 3.4 Hz, 15-H<sub>B</sub>), 3.42 (3H, s, -COOMe), 1.43 (3H, s, 16-Me), 0.89 (3H, s, 19-Me), 0.86 (3H, s, 18-Me), 0.79 (3H, s, 20-Me); δ<sub>C</sub> (CDCl<sub>3</sub>) See table; Anal. Calcd for C<sub>21</sub>H<sub>34</sub>O<sub>4</sub>: C<sub>1</sub>71.96; H, 9.78; Found C, 71.89; H, 9.74.

Acetylation of 4a: 5a: 474 mg of 4a dissolved in 1.5 ml of anhydrous pyridine were acetylated with 2 ml of  $Ac_2O$  at room temperature overnight. Ice was added and after one hour the product was recovered in the usual way, yielding 5a (438 mg). Colourless oil ;  $v_{max}$  (film) 1740, 1320, 1240, 1110, 960, 850 ;  $\delta_H$  (CDCl<sub>3</sub>), 6.32 (1H, m, 7-H), 4.19 (2H, m, 15-H), 4.02 (1H, m, 14-H), 3.30 (3H, s, -COOMe), 2.08 (3H, s, -AcO), 1.43 (3H, s, 16-Me), 0.88 (3H, s, 19-Me), 0.84 (3H, s, 18-Me), 0.78 (3H, s, 20-Me).

Hydrolysis of 5a: 6a, 7a.-HCl (1N, 8 ml) was added to a solution of 5a (438 mg) in acetone (22 ml). The mixure was stirred at room temperature for 2 days, and was then heated to  $40^{\circ}$  C for a further 3 hours. Following this, water was added and the solvent was eliminated under vacuum. The crude mixture was extracted with ether. The ethereal solution was washed with  $Na_2CO_3$  and water, dried ( $Na_2SO_4$ ) and concentrated in vacuo. The residue (364 mg) was dissolved in dry pyridine (1 ml) and acetylated with  $Ac_2O$  (1 ml) at room temperature for 16 hours. The excess of  $Ac_2O$  was hydrolyzed with ice-water and the product was extracted with ether in the usual way, giving 368 mg of a mixture of acetylderivatives that were separated by column chromatography on  $SiO_2$  and eluting with hexane; AcOET (95:5, gradient) to afford: 112 mg of the initial sustrate, 150 mg (41%) of 6a and 90 mg (24.5%) of 7a.

 $\frac{14 \text{ S.15-diacetoxy-7-labden-13 R.17-olide}}{14 \text{ S.15-diacetoxy-7-labden-13 R.17-olide}} : \underbrace{6a}_{\text{a.}}\text{-Colourless oil}; v_{\text{max}} \text{ (film ) 1750, 1710, 1630, 1240}; \delta_{\text{H}} \text{ (CDCl}_3) 7.02 \text{ (1H, m, 7-H), 5.24 (1H, dd, J = 8.3, 2.4 Hz, 14-H), 4.60 (1H, dd, J = 11.7, 8.3 Hz, 15-H<sub>A</sub>), 4.19 (1H, dd, J = 11.7, 2.4 Hz, 15-H<sub>B</sub>), 2.12 (3H, s, AcO-), 2.03 (3H, s, AcO-), 1.34 (3H, s, 16-Me), 0.91 (3H, s, 19-Me), 0.89 (3H, s, 18-Me), 0.77 (3H, s, 20-Me); \delta_{\text{C}} \text{ (CDCl}_3)}$  See table; Anal. Calcd for  $C_{24}H_{36}O_6: C$ , 68.53; H, 8.63; Found C, 68.48; H, 8.60.

Dehydration of 7a with POCl<sub>3</sub>/Py: 8a, 9a.-POCl<sub>3</sub> (freshly distilled, 0.75 ml) was added to a solution of 7a (86.8 mg) in anhydrous pyridine (8 ml) cooled to 0<sup>a</sup> C. Having completed the addition, the mixture was stirred at room temperature for 18 hours. Then ice and ether were added. The ethereal solution was washed with 2N HCl, aqueous 5% NaHCO<sub>3</sub> and water and dried over Na<sub>2</sub>SO<sub>4</sub>. The solvent was eliminated under vacuum, yielding a crude mixture (80 mg) which by p.t.l.c. on SiO<sub>2</sub> eluting twice with hexane: ether (50:50) afforded: 8a (31 mg) and 9a (29 mg).

Methyl-14 R,15-diacetoxy-7,13(16)-labdadien-17-oate; 8a. -Colourless oil;  $v_{max}$  (film) 3080, 1740, 1720, 1640, 1240 890;  $δ_{\rm H}$  (CDCl<sub>3</sub>) 6.68 (1H, m, 7-H), 5.38 (1H, dd, J = 7.8, 3.4 Hz, 14-H), 5.07 (1H, s, 16-H), 4.97 (1H, s, 16-H), 4.26 (1H, dd, J = 11.7, 3.4 Hz, 15-H<sub>A</sub>), 4.11 (1H, dd, J=11.7, 7.8 Hz, 15-H<sub>B</sub>), 3.71 (3H, s, -COOMe), 2.10 (3H, s, AcO-), 2.06 (3H, s, AcO-), 0.91 (3H, s, 19-Me), 0.88 (3H, s, 18-Me), 0.84 (3H, s, 20-Me);  $δ_{\rm C}$  (CDCl<sub>3</sub>) See table .

Methyl-14 R,15-diacetoxy-7,12-labdadien-17-oate: 9a.-Colourless oil;  $v_{max}$  (film) 1740, 1720, 1640, 1240;  $\delta_H$  (CDCl<sub>3</sub>) 6.71 (1H, m, 7-H), 5.49 (1H, t, J=5.8 Hz, 12-H), 5.29 (1H, dd, J=4.3, 4.3 Hz, 14-H), 4.17 (1H, dd, J=11.7, 8.3 Hz, 15-H<sub>A</sub>), 4.06 (1H, dd, J=11.7, 8.3 Hz, 15-H<sub>B</sub>), 3.67 (3H, s, -COOMe), 2.07 (3H, s, AcO-), 2.05 (3H, s, AcO-), 1.63 (3H, s, 16-Me), 0.91 (3H, s, 19-Me), 0.88 (3H, s, 18-Me), 0.80 (3H, s, 20-Me).

Alkaline hydrolysis of 8a: 1.-26 mg of 8a are saponificated at room temperature with 3 ml of 1M NaOH/MeOH for 6 hours. The usual procedure is followed, yielding 23 mg of 1.

Methyl-14 R,15-dihydoxy-7.13(16)-labdadien-17-oate: 1.- Colourless oil;  $[\alpha]_D$  -47.4° (c 0.4, CHCl<sub>3</sub>); CD  $\Delta\epsilon_{313}$  +1.2 (n-hexane);  $\nu_{max}$  (film) 3440, 1730, 1650;  $\delta_H$  (CDCl<sub>3</sub>) 6.71 (1H, m, 7-H), 5.08 (1H, s, 16-H), 4.94 (1H, s, 16-H), 4.18 (1H, dd, J=7.3, 3.9 Hz, 14-H), 3.73 (1H, dd, J=11.2, 3.9 Hz, 15-H<sub>A</sub>), 3.71 (3H, s, COOMe), 3.56 (1H, dd, J=11.2, 7.3 Hz, 15-H<sub>B</sub>), 0.91 (3H, s, 16-Me), 0.87 (3H, s, 18-Me), 0.83 (3H, s, 20-Me).  $\delta_C$  (CDCl<sub>3</sub>) See table; EIMS 70 ev, m/z (rel. int.):350[M\*](5), 332(15), 319(18), 287(30), 248(70), 190(30), 149(60), 109(100); Anal. Calcd for C<sub>21</sub>H<sub>34</sub>O<sub>4</sub>: C, 71.96; H, 9.78; Found C, 71.87; H, 9.72.

Sharpless reaction of 3: 4b.-A 100 ml, flask equipped with a Teflon-coated magnetic stirring bar is over-dried, then flushed with nitrogen. The flask is charged with dry dichloromethane (35 ml) and cooled by stirring in a -23 $^{12}$  C bath. The following liquids are then added sequentially via syringe while stirring in the cooling bath: Ti (iPrO)<sub>4</sub> (1.6 ml, 5.5 mmol,); D-(-)-DET (1.0 ml, 5.5 mmol), the mixture is stirred 5 min. before the next addition, 3 (1.84 g, 5.5 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) and, finally, t-butyl hydroperoxide (3.7 ml, 11 mmol). The homogeneous solution is then stored overnight (18 hours) in the freezer at -20 $^{12}$  C in the sealed reaction vessel. In the same manner as described for the previous Sharpless reaction, a crude mixture was obtained that was chromatographed on SiO<sub>2</sub>, to afford 4b (1,2 g, 65%). Colourless oil;  $v_{max}$  (film) 3450, 3050, 1645, 1270, 1250, 1190, 975, 850;  $\delta_{H}$  (CDCl<sub>3</sub>) 6.22 (1H, m, 7-H), 4.04 (1H, dd, J=7.8, 3.4 Hz, 14-H), 3.94 (1H, dd, J=11.2, 7.8 Hz, 15-H<sub>A</sub>), 3.62 (1H, dd, J=11.2, 3.4 Hz, 15-H<sub>B</sub>), 3.38 (3H, s, COOMe), 1.36 (3H, s, 16-Me), 0.88 (3H, s, 19-Me) 0.84 (3H, s, 18-Me), 0,74 (3H, s, 20-Me).  $\delta_{C}$  (CDCl<sub>3</sub>) See table; Anal. Calcd for C<sub>21</sub>H<sub>34</sub>O<sub>4</sub>: C, 71.96; H, 9.78; Found C, 71.94; H, 9.80.

Acetylation of 4b: 5b.-1.20 g of 4b dissolved in 2 ml of anhydrous pyridine were acetylated with 2 ml (20.6 mmol) of  $Ac_2O$  at room temperature overnight. Ice was added and after 1 hour product was recovered in the usual way, yielding 5b (1.20 g). Colourless oil;  $v_{max}$  (film) 1740, 1650, 1320, 1240, 1110, 960, 850.;  $\delta_H$  (CDCl<sub>3</sub>) 6.23 (1H, m, 7-H), 4.29 (1H, dd, J=11.7, 7.8 Hz, 15-H<sub>A</sub>), 4.18 (1H, dd, J=11.7, 3.9 Hz, 15-H<sub>B</sub>), 4.07 (1H, dd, J=7.8, 3.9 Hz, 14-H) 3.37 (3H, s, -COOMe), 2.08 (3H, s, AcO-), 1.37 (3H, s, 16-Me), 0.87 (3H, s, 19-Me), 0.83 (3H, s, 18-Me), 0.73 (3H, s, 20-Me).  $\delta_C$  (CDCl<sub>3</sub>) See table .

Hvdrolysis of 5b: 6b, 7b.- 250 mg of 5b were dissolved in MeOH / H<sub>2</sub>O (80:20). pH was adjusted to 1 by addition of 2M HCl. The mixture was stirred for 36 hours at 45° C. Following this, water was added and the solvent was eliminated under vacuum. The crude mixture was extracted with ether and the usual procedure was followed, a crude mixture was obtained that was dissolved in dry pyridine (2 ml) and then acetylated with Ac<sub>2</sub>O (2 ml) at room temperature overnight. Ice was added and after one hour the product was recovered in the usual way, yielding after chromatography on SiO<sub>2</sub> and elution with hexane:AcOEt (90:10, gradient): 5b (108 mg, 43.4%), 6b (443 mg, 17.3%) and 7b (46 mg, 18,7%).

 $\frac{14 \text{ R,}15\text{-diacetoxy-}7\text{-labden-}13\text{S,}17\text{-olide}}{13\text{C,}17\text{-olide}} : \frac{6\text{b}}{6\text{L}} \text{ .-Colourless oil; } v_{\text{max}} \text{ (film); } 1760, 1710, 1640, 1465, 1375, 980, 860; } \delta_{\text{H}} \text{ (CDCl}_3) 7.12 \text{ (1H, m, 7-H), } 5.11 \text{ (1H, dd, J=8.3, 2.4 Hz, 14-H), 4.55 (1H, dd, J=12.2, 2.4 Hz, 15-H_A), 4.18 (1H, dd, J=12.2, 8.3 Hz, 15-H_B), 2.09 \text{ (3H, s, AcO-), } 2.02 \text{ (3H, s, AcO-), } 1.48 \text{ (3H, s, 16-Me), } 0.88 \text{ (6H, s, 19-Me, 18-Me), } 0.71 \text{ (3H, s, 20-Me). } \delta_{\text{C}} \text{ (CDCl}_3) \text{ See table, Anal. Calcd for } C_{24}H_{36}O_6 : C, 68.53; H, 8.63; \text{ Found C, } 68.56; H, 8.58.}$ 

Methyl-14 R,15-diacetoxy-13 S-hydroxy-7-labden-17-oate: 7b.-Colourless oil;  $[\alpha]_D$ -21.8¢ (c 2.7, CHCl<sub>3</sub>);  $\nu_{max}$  (film) 1750, 1720, 1650, 1470, 1250, 870;  $\delta_H$  (CDCl<sub>3</sub>) 6.72 (1H, m, 7-H), 5.06 (1H, dd, J=8.3, 2.4 Hz, 14-H), 4.52 (1H, dd, J=12.2, 2.4 Hz, 15-H<sub>A</sub>), 4.17 (1H, dd, J=12.2, 8.3 Hz, 15-H<sub>B</sub>), 3.66 (3H, s, -COOMe), 2.09 (3H, s, Aco-), 2.01 (3H, s, AcO-), 1.19 (3H, s, 16-Me), 0.89 (3H, s, 19-Me), 0.85 (3H, s, 18-Me), 0.81 (3H, s, 20-Me);  $\delta_C$  (CDCl<sub>3</sub>) See table.

Hydrolysis of 6b; 7b.-85 mg of 6b were saponificated with 10 ml of 10%  $K_2CO_3$ /MeOH with a few drops of water. The mixture was kept refluxing overnight. Following this, ICH<sub>3</sub> was added and refluxing again for 18 hours. The solvent was eliminated and CH<sub>2</sub>Cl<sub>2</sub> was added. The organic solution was washed with Na<sub>2</sub>CO<sub>3</sub> and water, dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. The residue (64 mg) was dissolved in dry pyridine (1 ml) and was acetylated with Ac<sub>2</sub>O (1 ml) at room temperature overnight. The excess of Ac<sub>2</sub>O was hydrolyzed with ice-water and the product was extracted with ether in the usual way, giving 7b.

Dehydration of 7b with POCl<sub>3</sub>/Py\_.-POCl<sub>3</sub> (freshly distilled, 0.75 ml) was added to a solution of 7b (86 mg) in anhydrous pyridine (7 ml), cooled to 0° C under N<sub>2</sub>. Having completed the addition, the mixture was stirred at room temperature for 24 hours. In the same manner as described for the previous dehydration, a crude mixture (70 mg) was obtained that by p.t.l.c. on SiO<sub>2</sub>, eluting twice with hexane:AcOEt (60:40) afforded: 30 mg of 8b and 30 mg of 9b.

Methyl-14 S.15-diacetoxy-7.13(16)-labdadien-17-oate: 8b.-Colourless oil;  $v_{max}$  (film) 3080, 1750, 1720, 1650, 1460, 1440, 1370, 1240, 1050, 950, 890;  $\delta_{H}$  (CDCl<sub>3</sub>) 6.68 (1H, m, 7-H), 5.38 (1H, dd, J=7.8, 3.4 Hz, 14-H), 5.06 (1H, s, 16-H), 4.96 (1H, s, 16-H), 4.24 (1H, dd, J=12.2, 3.4 Hz, 15-H<sub>A</sub>), 4.12 (1H, dd, J=12.2, 7.8 Hz, 15-H<sub>B</sub>), 3.70 (3H, s, -COOMe), 2.09 (3H, s, AcO-), 2.05 (3H, s, AcO-), 0.90 (3H, s, 19-Me), 0.87 (3H, s, 18-Me), 0.82 (3H, s, 20-Me).  $\delta_{C}$  (CDCl<sub>2</sub>) See table.

Methyl-14 S.15-diacetoxy-7,12-labdadien-17-oate :9b.-Colourless oil;  $v_{max}$  (film) 1750, 1715, 1650, 1440, 1370, 1220, 960;  $\delta_{H}$  (CDCl<sub>3</sub>) 6.72 (1H, m, 7-H) 5.49 (1H, t, J=6.3 Hz, 12-H), 5.29 (1H, dd, J=7.8, 4.4 Hz, 14-H), 4.15 (1H, dd, J=11.2, 4.4 Hz, 15-H<sub>A</sub>), 4.07 (1H, dd, J=11.2, 7.8 Hz, 15-H<sub>B</sub>), 3.65 (3H, s, -COOMe), 2.06 (3H, s, AcO-), 2.04 (3H, s, AcO-), 1.62 (3H, s, 16-Me), 0.91 (3H, s, 19-Me), 0.87 (3H, s, 18-Me), 0.81 (3H, s, 20-Me). $\delta_{C}$  (CDCl<sub>3</sub>) See table.

Alkaline hydrolysis of 8b:2.-20 mg of 8b were saponificated at room temperature with 3 ml of 1M NaOH/MeOH for 7 hours. Following this, the solvent was eliminated under vacuum and the product was extracted with ether, in the usual way, giving 15 mg of 2.

Methyl-14 S.15-dihydroxy-7,13(16)-labdadien-17-oate : 2.- Colourless oil;  $[\alpha l_D$  -37.1° (c 0.3 CHCl<sub>3</sub>); CD  $\Delta \epsilon_{313}$  - 0.3 (n-hexane);  $v_{max}$  (film) 3400, 1720, 1650, 1450, 1240, 900;  $\delta_H$  (CDCl<sub>3</sub>) 6.70 (1H, m 7-H), 5.09 (1H, s, 16-H), 4.94 (1H, s, 16-H), 4.20 (1H, dd, J=8.9, 7.3 Hz, 14-H), 3.70 (3H, s, -COOMe), 3.69 (1H, dd, J=11.2, 8.9 Hz, 15-H<sub>A</sub>), 3.57 (1H, dd, J=11.2, 7.3 Hz, 15-H<sub>B</sub>), 0.91 (3H, s, 19-Me), 0.87 (3H, s, 18-Me), 0.82 (3H, s, 20-Me).  $\delta_C$  (CDCl<sub>3</sub>) See table; EIMS 70 ev, m/z (rel. int.):350[M<sup>+</sup>](3), 332(12), 319(120, 287(28), 248(65), 190(45), 149(74), 109(100); Anal. Calcd for  $C_{21}H_{34}O_4$ : C,71.96; H, 9.78; Found C, 71.92; H, 9.73.

Reaction of 3 with m-CPBA: 10a/10b.-Compound 3 (70 mg, 0.21 mmol) was dissolved in dry  $CH_2Cl_2$  (5 ml) and m-CPBA (41 mg, 0.24 mmol) was added. The mixture was stirred at room temperature for 4 hours. The solvent was eliminated and ether was added. The ethereal solution was washed with  $Na_2CO_3$  and brine, dried ( $Na_2SO_4$ ) and concentrated in vacuo, to obtain 68 mg of 10a/10b.-Colourless oil;  $v_{max}$  (film) 3450, 1730, 1655, 1470, 1440, 1280, 1250, 1150, 1070, 790;  $\delta_H$  (CDCl<sub>3</sub>) 6.68 (1H, m), 3.74 (2H, m), 3.71 (3H, s, -COOMe), 2.95 (1H, m, 14-H), 1.27 (3H, s), 0.90 (3H, s), 0.86 (3H, s), 0.82 (3H, s).

Acknowledgements. The authors thank the Comision Asesora de Investigacion Cientifica y Tecnica for financial support (PB 85 0300).

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